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Note

High-performance liquid chromatographic separation of amino sugars and peptides with metal ion-modified mobile phases

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The use of impregnated supports for the separation of closely related compounds has been reported for thin-layer chromatographic (TLC) systems^{1,2}. In an application of these TLC techniques to high-performance liquid chromatography, Schomburg and Zegarski³ have reported the use of an argentated mobile phase with a reversed-phase partition system for the separation of geometrical isomers of 2-alkenes, oleic and elaidic acid methyl esters. The separation of pharmaceutical compounds using a mobile phase containing silver nitrate has been reported by Tscherne and Capitano⁴. Other metal ions such as Ni(II), Cd(II) and Zn(II) have been used in two different manners in the reversed-phase mode. The metal ion can be added to the mobile phase directly⁵ or in a chelated form⁶.

In the present study we have used cadmium sulphate and zinc acetate in the mobile phase for separation of amino sugars and some peptides on an amino column, and explain the probable mechanism.

EXPERIMENTAL

The HPLC apparatus consisted of a Tracor 990 pump, a Rheodyne 7125 injector and a Schoeffel SF 770 variable-wavelength UV detector monitoring at 180–220 nm. 600-NH amino columns (Alltech) were used for all the studies. A precolumn of silica gel (Universal Scientific) was used to improve the life and stability of the column. The amino sugars and peptides were of reagent grade (Sigma, St. Louis, MO, U.S.A.). The CdSO₄, zinc acetate and acetonitrile (HPLC Grade) were obtained from J. T. Baker (Phillipsburg, NJ, U.S.A.). 2,2,2-Trifluoroethanol (sequanol grade) was purchased from Pierce (Rockford, IL, U.S.A.).

All chromatograms were developed at room temperature. Amino sugars and peptides were dissolved in water. The columns were allowed to equilibrate in the appropriate mobile phase for about 120 ml prior to chromatography. The UV absorbance of the amino sugars was monitored at 190 nm and that of peptides at 202 nm.

RESULTS AND DISCUSSION

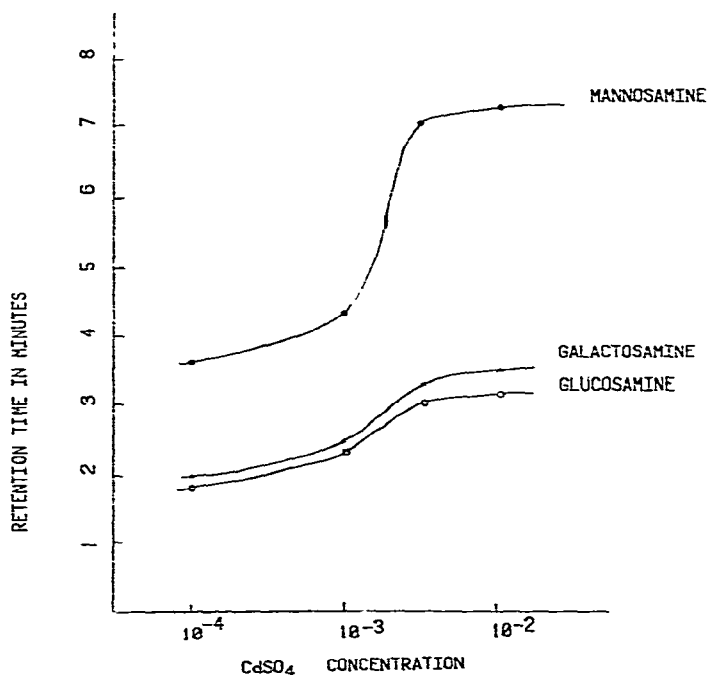
We first measured the effect of CdSO₄ on the retention of amino sugars. Although there is no retention of amino sugars with pure water as the mobile phase,

TABLE I

EFFECT OF CdSO_4 CONCENTRATION IN THE MOBILE PHASE ON THE RETENTION TIMES (min) OF AMINO SUGARS

Amino sugars	CdSO_4 concentration in water			
	$10^{-4} M$	$10^{-3} M$	$5 \times 10^{-3} M$	$10^{-2} M$
Glucosamine	1.8	2.5	3.0	3.2
Galactosamine	2.0	2.4	3.3	3.5
Mannosamine	3.6	4.3	7.15	7.3

the retention times increased with the amount of CdSO_4 in the mobile phase (Table I). Fig. 1 shows that the retention times increase in a sigmoidal fashion to a maximum at $10^{-2} M$. The peak width also increased with increasing CdSO_4 concentration, an effect which could be reduced by addition of the organic modifiers acetonitrile and 2,2,2-trifluoroethanol to the mobile phase. It was found that $10^{-3} M$ cadmium sulphate-acetonitrile-trifluoroethanol (83:15:2) as mobile phase gave good separation of all amino sugars (see Fig. 2). We observed two peaks for each amino sugar with this mobile phase, one of the peaks having the same retention time for all the amino sugars. Since the samples are amino sugar hydrochlorides, one of the peaks is presumably due to HCl and the other is due to the amino sugar. In further support for this interpretation, we observed that injection of HCl gives a peak having the same retention time as the peak which is common in all the amino sugars. Hence this

Fig. 1. Influence of CdSO_4 concentration on retention of amino sugars. Mobile phase: CdSO_4 in water.

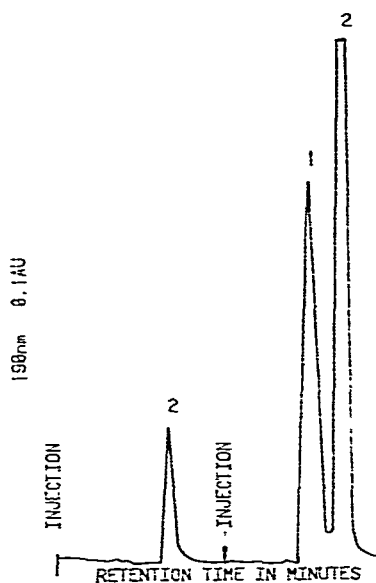


Fig. 2. Separation of glucosamine (1) and HCl (2). Conditions: eluent, 83 ml 10^{-3} M CdSO_4 + 15 ml acetonitrile + 2 ml trifluoroethanol; flow-rate, 1.0 ml/min.

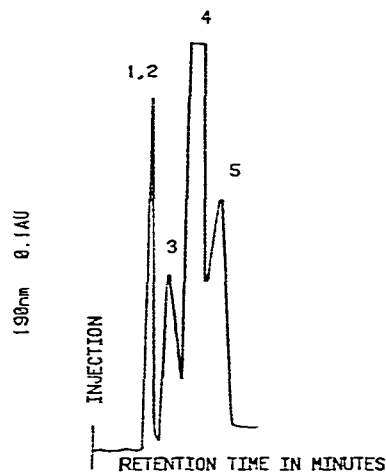


Fig. 3. Separation of sugars. Eluent as in Fig. 2. Solutes: 1 = glucose; 2 = galactose; 3 = glucosamine; 4 = HCl; 5 = mannosamine.

system could be useful for the isolation of the free base from its hydrochloride. Fig. 2 shows the chromatogram of the HCl peak alone and that of glucosamine hydrochloride.

Fig. 3 shows that this mobile phase may be used to separate neutral sugars from amino sugars. The latter are all retained under these conditions, while the former migrate with the injection peak. We believe that the retention of the amino sugars results from the formation of weak metal ion complexes with Cd^{2+} forming bridges between the stationary phase amine and that of the amino sugar. The retention times of three different amino sugars are summarized in Table II.

Influence of CdSO_4 on the retention of peptides

Although the influence of CdSO_4 on the retention of peptides is similar to that

TABLE II

HPLC SEPARATION OF NEUTRAL SUGARS FROM AMINO SUGARS

Eluent: 83 ml 10^{-3} M CdSO_4 + 15 ml acetonitrile + 2 ml trifluoroethanol. Flow-rate: 1 ml/min.

Compounds	Retention time (min)
Glucose, galactose, mannose, maltose, lactose	1.4
HCl	2.6
Glucosamine	1.7
Galactosamine	2.0
Mannosamine	3.2

TABLE III

HPLC SEPARATION OF SOME PEPTIDES

Eluent: 89 ml $5 \cdot 10^{-4}$ M CdSO_4 + 10 ml acetonitrile + 1 ml trifluoroethanol. Flow-rate: 1.5 ml/min. Detection: 202 nm. pK_2 values are taken from ref. 7.

Peptide	Retention time (min)	pK_2 (NH_3) ⁺
Void volume	1.0	
Val-Val	3.2	
Ala-Val	4.1	
Ala-Leu	4.5	
Ala-Ala	5.2	8.14
Ala-Gly	5.4	8.18
Gly-Leu	9.0	8.29
Gly-Thr	4.0	8.40
Gly-Gly	11.1	8.17
Gly-Gly-Gly	13.7	7.91
Gly-Gly-Gly-Gly	15.2	7.75
Gly-Gly-Gly-Gly-Gly-Gly	15.8	7.60

on the amino sugars, *i.e.*, the retention increases with increasing CdSO_4 concentration in the mobile phase, the peptides are retained more strongly than amino sugars for the same concentration of CdSO_4 . The most suitable mobile phase for the separation of peptides was found to be $5 \cdot 10^{-4}$ M cadmium sulphate-acetonitrile-tri-

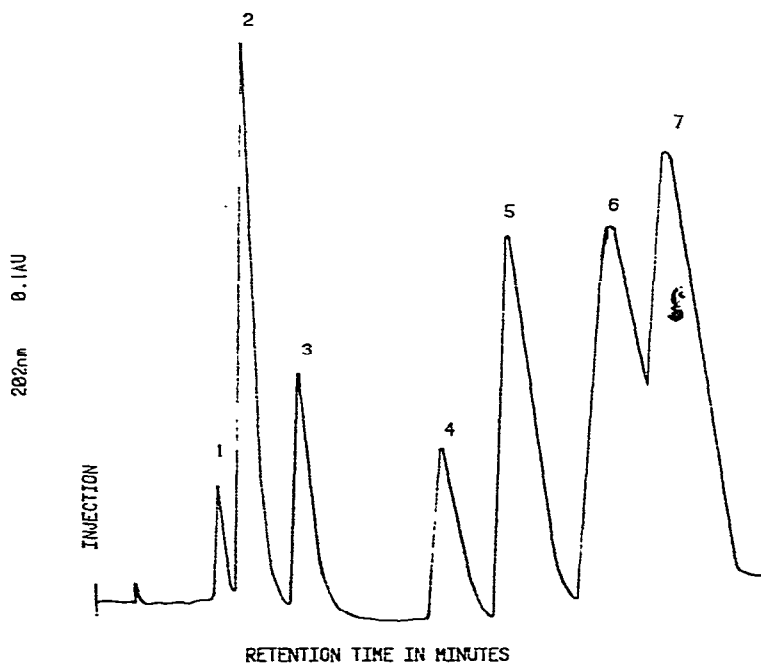


Fig. 4. Separation of peptides. Eluent: 89 ml $5 \cdot 10^{-4}$ M CdSO_4 + 10 ml acetonitrile + 1 ml trifluoroethanol. Solutes: 1 = Val-Val; 2 = Gly-Thr; 3 = Ala-Gly; 4 = Gly-Leu; 5 = Gly-Gly; 6 = Gly-Gly-Gly; 7 = Gly-Gly-Gly-Gly.

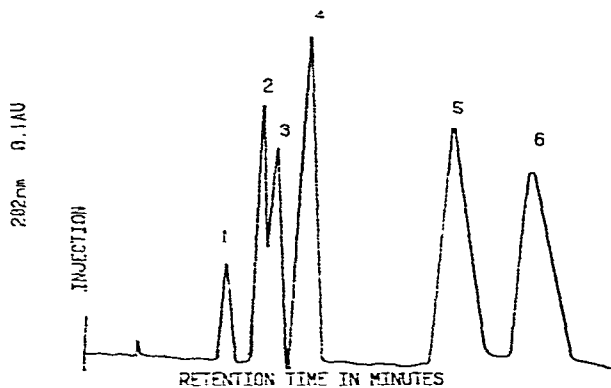


Fig. 5. Separation of peptides. Eluent as in Fig. 4. Solutes: 1 = Val-Val; 2 = Ala-Val; 3 = Ala-Leu; 4 = Ala-Ala; 5 = Gly-Leu; 6 = Gly-Gly.

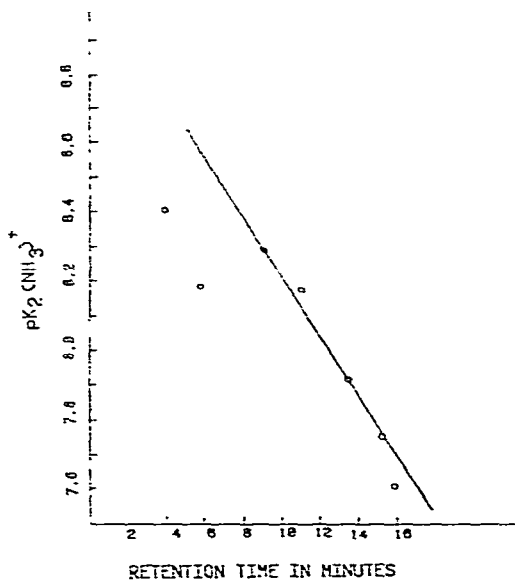
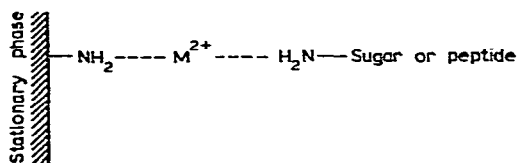


Fig. 6. Relationship between pK_2 (NH_3^+) of peptides and their retention times.

fluoroethanol (89:10:1). The retention times are shown in Table III and the separation of some of the peptides is given in Figs. 4 and 5.

In order to establish whether there is any correlation between the retention times and basicities of the peptides, we plotted pK_2 value of peptide vs. retention time (Fig. 6). It can be seen that the retention decreases as the pK_2 value increases. Similar behavior was observed by Yasuda¹ in the TLC separation of aromatic amines on silica gel layers impregnated with cadmium sulphate. Hence the chromatographic behavior of amino sugars and peptides seems to be mainly due to the weak complex formation between the metal ion and the NH_2 group of the sugars or peptides as shown below:



We next attempted to separate amino sugars from peptides using several mobile phases containing CdSO_4 . Since both the amino sugars and peptides were retained, we could not separate amino sugars from peptides. So we tried zinc acetate in the mobile phase which forms very weak complexes with amino sugars but stronger complexes with peptides. With 10^{-3} M zinc acetate as the mobile phase the amino sugars are not retained and elute with the solvent front while the peptides are retained. The results are shown in Table IV. Hence this system is most suitable for the separation of amino sugars from peptides.

TABLE IV

HPLC SEPARATION OF AMINO SUGARS FROM PEPTIDES

Eluent: 10^{-3} M zinc acetate. Flow-rate: 1.5 ml/min. Column: amino column.

Compound	Retention time (min)
<i>Sugars</i>	
Glucosamine	2.2
Galactosamine	2.2
Mannosamine	2.2
Glucose	2.2
Galactose	2.2
<i>Peptides</i>	
Ala-Ala	10.0
Ala-Leu	10.2
Ala-Val	8.4
Ala-Gly	9.3
Gly-Gly	11.0
Gly-Gly-Gly	13.6
Gly-Gly-Gly-Gly	14.2
Gly-Leu	12.3
Gly-Thr	10.2
Val-Val	8.9

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